Studies in Spiroheterocycles. Part XVII.

Synthesis of Novel Fluorine Containing Spiro[indole-pyranobenzopyran] and Spiro[indenopyran-indole] Derivatives

Krishna C. Joshi*, Anshu Dandia, Sunita Baweja and Amitabh Joshi

Department of Chemistry, University of Rajasthan, Jaipur-302004, India Received October 3, 1988

An elegant one-step synthesis of two novel spiro ring systems viz: spiro[3H-indole-3,4'-(2'-amino-3'-carbonitrile-[4'H]-pyrano[3,2-c]benzopyran)]-2,5'(1H)-diones and spiro[(2-amino-3-carbonitrile-indeno[1,2-b]pyran)-4(5H),3'-[3H]indole]-2',5(1'H)-diones in 80-85% yields is described. The spiro heterocycles were prepared by the reactions of fluorine containing 3-dicyanomethylene-2H-indol-2-ones with 4-hydroxy-2H-1-benzopyran-2-one and 1H-indene-1,3(2H)-dione respectively. The synthesized compounds have been characterized on the basis of elemental analyses, ir, pmr, 1°F nmr and mass spectral data.

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As a part of our continuing interest on the synthesis of new fluorine containing heterocycles incorporating spiro indolines [1] for possible pharmacological evaluation, we now wish to report the synthesis of a novel ring system, viz: spiro[3H-indole-3,4'-(2'-amino-3'-carbonitrile-[4'H]-pyrano[3,2-c]benzopyran)]-2,5'(1H)-dione (2) by the reaction of fluorine containing 3-dicyanomethylene-2H-indol-2-ones 1 with 4-hydroxy-2H-1-benzopyran-2-one. Further, we report another new facile approach to the synthesis of a seccond spiro system, viz: spiro[(2-amino-3-carbonitrile-indeno[1,2-b]pyran)-4(5H),3'-[3H]indole]-2',5(1'H)-dione (3) by the reaction of 1 with 1H-indene-1,3(2H)-dione.

The medicinal application of spiro[indolo-pyrans] as muscle relaxants and antiinflammatory agents is well known [2,3]. Further, the bioactivity of coumarins [4,5] and

the indene derivatives [6] is well established. Recently Michael adduct of arylidene indenedione with coumarins were reported as active anticoagulant [7]. Nevertheless, no report has been cited on spiro indolines incorporating pyranobenzopyrans and with indenopyrans, which may lead to the production of compounds with altered/enhanced bioactivity. Besides, practically no work has been done on fluorine containing spiro[indolo-pyrans]. The coumarin and indene derivatives have, however, been reported to undergo Michael addition reaction [8,9].

In view of these observations, we have now investigated the reaction of 1 with 4-hydroxy-2H-1-benzopyran-2-one to explore the possibility of the formation of a novel spiro system incorporating indole and pyranobenzopyran moieties. In yet another attempt to obtain a different spiro system, we explored the reaction of 1 with 1H-indene-

Michael adduct

Scheme 1

Table I

Characteristic Physical and Analytical Data of Spiro[indole-pyranobenzopyran] and spiro[indenopyran-indole] Derivatives

Compound No.	R	R¹	MP °C	Yield %	Formula	Analysis %					
						С	H Calcd.	N	С	H Found	N
2a	Н	Н	291	86	C ₂₀ H ₁₁ N ₂ O ₄	67.22	3.08	11.76	67.28	3.12	11.72
2 b	COCH ₃	Н	276	88	$C_{22}H_{13}N_3O_5$	66.16	3.25	10.52	66.12	3.28	10.56
2c	H	5-F	321	84	C ₂₀ H ₁₀ FN ₈ O ₄	64.00	2.66	11.20	64.08	2.62	11.24
2d	H	6-F	279	85	CaoH10FN8O4	64.00	2.66	11.20	64.05	2.60	11.18
3a	Н	H	242	87	$C_{so}H_{11}N_sO_s$	70.38	3.22	12.31	70.35	3.25	12.34
3b	CH,	Н	236	64	$C_{s_1}H_{1s}N_sO_s$	71.38	3.68	11.89	71.40	3.65	11.84
3 c	Н	5-F	249	86	C ₂₀ H ₁₀ FN ₈ O ₈	66.85	2.78	11.69	66.82	2.76	11.66
3d	Н	6-F	210	90	C _{so} H ₁₀ FN _s O _s	66.85	2.78	11.69	66.88	2.80	11.71

1,3(2H)-dione to afford spiro[indenopyran-indole]. There is only one report in the literature mentioning the synthesis of spiro[indenopyran-indole], but this is through a complicated route involving the cycloaddition of 2-(2-oxindolin-3-ylidene)-1,3-indenedione with ethyl vinyl ether and gives comparatively low yields [10].

The reaction of an appropriate indole-2,3-dione with malononitrile in absolute ethanol, in presence of piperidine, results in the formation of 1,3-dihydro-3-dicyanomethylene-2*H*-indol-2-ones 1. The Michael reaction of 1, having an electron attracting group on the exomethylene carbon is interesting, as it can afford either a Michael adduct which can exist as such or this adduct can be enolized and converted into the spiro-pyran system of C-3 of oxindole (Scheme 1).

The Michael reaction of violet coloured 1 with 4-hydroxy-2H-1-benzopyran-2-one and 1H-indene-1,3(2H)dione afforded light yellow coloured compounds. The ir spectra of these compounds displayed characteristic absorption bands at 3380 cm⁻¹ corresponding to free NH₂ group and at 3120 cm⁻¹ due to NH stretching. Strong absorption bands corresponding to conjugated C=N and two carbonyl groups were observed at 2200-2190, 1700 and 1680 cm⁻¹ respectively. The pyran ether linkage was observed at 1180 cm⁻¹. Formation of light coloured compounds and disappearance of exocyclic C=C absorption at 1620 cm⁻¹ indicated the relief in the conjugation of oxindolidene system of 1. The pmr spectra displayed aromatic protons in the region δ 6.8-7.4 and a characteristic broad signal at 7.7-7.9 ppm corresponding to NH₂ protons which disappeared on deuteration. An inspection of ir and pmr provides no evidence for the Michael adduct as characteristic C-H stretching and hydroxyl absorption in ir and two singlet, corresponding to methine proton and OH group in pmr spectra should have been observed. On the basis of these observations, along with the spectral data,

the products formed with 4-hydroxy-2H-1-benzopyran have been identified as spiro[indole-pyranobenzopyran] derivatives **2a-d** and with 1*H*-indene-1,3(2*H*)-dione as spiro[indenopyran-indole] derivatives **3a-d**.

Formation of spiro compounds 2a-d and 3a-d has further been confirmed on the basis of mass spectra. In the mass spectrum of compound 2c, the molecular ion peak was observed at m/z 375 (100%) and in 3c at m/z 359 (100%) corresponding to their molecular weights.

The presence and position of fluorine was also confirmed by ¹⁹F nmr. Fluorine attached at the 5- and 6-position of indole ring was observed at δ -115 and -112 ppm respectively.

EXPERIMENTAL

Melting points are uncorrected. Infrared spectra were recorded using Perkin-Elmer 557 spectrophotometer. Pmr spectra were recorded on Jeol (model FX 90Q) at 89.55 MHz using TMS as external reference and DMSO-d₆ or TFA as solvents. ¹⁹F nmr spectra were recorded on Jeol (model FX 90Q) at 84.25 MHz and taken in DMSO-d₆ using hexafluorobenzene (at δ –162.9 ppm) as external reference. The mass spectra were recorded on MS-30 and MS-50 Kratos mass spectrometer operating at an ionisation potential of 70 eV.

1,3-Dihydro-3-dicyanomethylene-2*H*-indol-2-ones 1.

These compounds were prepared following the method of Yokayama [11] using appropriate indole-2,3-dione and malononitrile. However, the 1-methyl derivative was obtained at room temperature on mixing the 1-methylindole-2,3-dione and malononitrile in absolute ethanol even in the absence of a catalyst.

1-Methyl-1,3-dihydro-3-dicyanomethylene-2H-indol-2-one.

This compound was obtained in 74% yield, mp 214°.

Spiro[3H-indole-3,4'-(2'-amino-3'-carbonitrile-[4'H]-pyrano[3,2-c]-benzopyran]-2,5'(1H)-diones **2a-d**.

A mixture of appropriate 3-dicyanomethylene-2H-indol-2-one (1) (0.01 mole) and 4-hydroxy-2H-1-benzopyran-2-one (0.01 mole)

in absolute ethanol (40 ml) was treated with two drops of piperidine as a catalyst and the mixture was stirred at room temperature. After two hours, a light yellow coloured compound separated which was filtered and recrystallized from ethanol.

The 1-acetyl derivative 2b was obtained after stirring for 30 minutes.

Spiro[(2-amino-3-carbonitrile-indeno[1,2-b]pyran)-4(5H),3'-[3H]-indole]-2',5(1'H)-diones 3a-d.

These compounds were prepared following the same procedure as above, using appropriate 1 and 1*H*-indene-1,3(2*H*)-dione.

The physical and analytical data of synthesized spiro compounds 2a-d and 3a-d are recorded in Table I.

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